М1.	(a)	neutrons and protons				
	(b)	0	1			
		(+)1	1			
	(c)	(i) total positive charge = total negative charge  accept protons and electrons have an equal opposite charge	1			
		(because) no of protons = no of electrons	1			
		(ii) ion	1			
		positive	1			
	(d)	Marks awarded for this answer will be determined by the quality of communication as well as the standard of the scientific response. Examiners should apply a best-fit approach to the marking.				
		0 marks No relevant content				
		Level 1 (1 – 2 marks) There is a basic description of at least <b>one</b> of the particles in terms of its characteristics.				

Level 2 (3 – 4 marks)

There is a clear description of the characteristics of **both** particles **or** 

a full description of either alpha **or** beta particles in terms of their characteristics.

## **Level 3 (5 – 6 marks)**

There is a clear and detailed description of **both** alpha and beta particles in terms of their characteristics.

#### examples of the physics points made in the response:

#### structure

- alpha particle consists of a helium nucleus
- alpha particle consists of 2 protons and 2 neutrons
- a beta particle is an electron
- a beta particle comes from the nucleus

#### penetration

- alpha particles are very poorly penetrating
- alpha particles can penetrate a few cm in air
- alpha particles are absorbed by skin
- alpha particles are absorbed by thin paper
- beta particles can penetrate several metres of air
- beta particles can pass through thin metal plate / foil
- beta particles can travel further than alpha particles in air
- beta particles can travel further than alpha particles in materials eg metals

#### deflection

- alpha particles and beta particles are deflected in opposite directions in an electric field
- beta particles are deflected more than alpha particles
- alpha particles have a greater charge than beta particles but beta particles have much less mass

or

beta particles have a greater specific charge than alpha particles

6

# M2.(a) (i) all correct

accept presented as a tally chart

Number of protons	3
Number of electrons	3
Number of neutrons	4

allow 1 mark for 1 correct

2

(ii) 7

reason may score even if 7 not chosen

1

number of protons and neutrons

accept number of particles in the nucleus

accept number of nucleons

do **not** accept number of electrons and neutrons

1

(b) an ion

1

(c) (i) smaller than

1

(ii) radon loses an alpha (particle)

or

radon loses an (alpha) particle

or

(mass of) polonium plus an alpha = (mass) radon

or

radon loses 2 protons and 2 neutrons (to become polonium)

accept radon has less protons and neutrons

[7]

1

М3.	(a)	proton			
		electron			
		neutron	all 3 in correct order allow 1 mark for 1 correct do not accept letters p, e, n	2	
	(b)	4	reason only scores if 4 is chosen	1	
		number of			
			accept number of electrons accept there are 4 protons and 4 electrons		
			do <b>not</b> accept there are 4 protons and electrons	1	
	(c)	The atom I	loses an electron.	1	[5]
M4.	(;	a) (i) <b>L</b>	-	1	
		(ii) <b>M</b>		1	
	(b)	To make a	smoke detector work.	1	
	(c)	40	no tolerance	1	[4]

**M5.** (a) electron(s)

1

(b) 3<sup>rd</sup> box ticked

The model cannot explain the results from a new experiment

1

(c) all three correct

Particle
Proton
Electron
Neutron

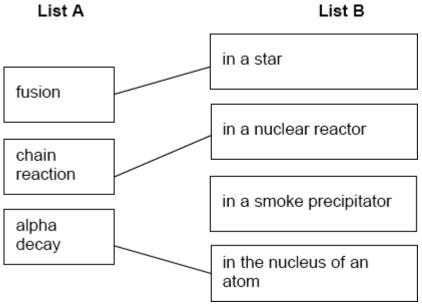
allow 1 mark for 1 correct

2

[4]

## **M6.three** lines correct

allow 1 mark for each correct line if more than 1 line is drawn from a box in **List A**, mark each line incorrect



[3]

M7.	(a)	(i)	neutron	1	
		(ii)	neutron proton both required, either order	1	
		(iii)	2	1	
			number of <u>protons</u> do not accept number of electrons	1	
	(b)	(i)	any <b>one</b> from:		
			• beta		
			gamma     accept correct symbols     accept positron / neutrino / neutron     cosmic rays is insufficient	1	
		(ii)	electrons	1	
		(iii)	are highly ionising	1	
	(c)	(i)	mutate / destroy / kill / damage / change / ionise  Harm is insufficient	1	
		(ii)	much smaller than	1	[9]

M	3. (	a)	(i)	half / 50 %	1		
		(ii)	Mea	sure the radon gas level in more homes in this area	1		
	(b)	(i)	86		1		
		(ii)	222		1		[4]
<b>M9</b> .(a)	protor	1		all 3 in correct order			
		elect		allow 1 mark for 1 correct do not			
		neut	ron	accept letters p, e, n		2	
	(b)	9		reason only scores if 9 is chosen		1	
		num	ber of	neutrons and protons		1	[4]

# **M10.**(a) neutron discovered

(b) neutron

all 3 in correct order

electron

allow 1 mark for 1 correct

proton

<sup>2</sup> [3]

1